

Hence, FIRST LAW of THERMODYNAMICS (applied to a closed system)

- The change in internal energy of a closed system is the energy that enters or leaves the system through boundaries as heat or work. i.e.
- $\Delta U = q + w$ 
  - $\Delta U = U_f - U_i$
  - $q$  – heat applied to system
  - $W$  – work done on system
  - When energy leaves the system,  $\Delta U = -ve$  i.e. decrease internal energy
- When energy enter the system,  $\Delta U = +ve$  i.e. added to internal energy

# Elements Of Applied Thermodynamics

**SJ Ball**



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