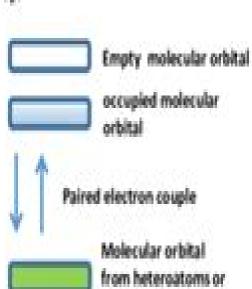


Molecules with alternating singles and double bonds, X = Hydrogen; Heteroatom; conjugated group, cyclic aromatic group. Y = Carbon; heteroatom, cyclic aromatic group.

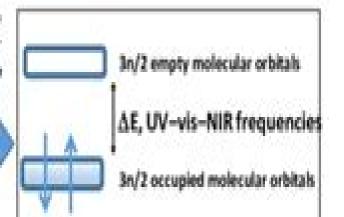


conjugated group

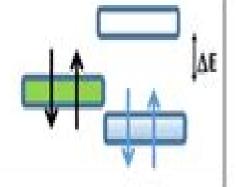


p orbitals along the backbone, each one contains one electron Overlapping of 3n atomic p orbitals generates 3n molecular orbitals, 3n/2 are occupied by 3n electrons, 3n/2 are empty

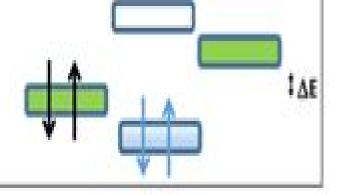
X, Y = Hydrogen, Carbon



X or Y = heteroatom: no bonding orbitals (green colored), containing electron couples, sit between the n occupied and n empty orbitals, lowering the ΔE.



X = conjugated group: orbitals containing electron couples and empty orbitals (green color) sit between the n occupied and n empty orbitals, lowering the ΔE.



Electronic Properties Of Conjugated Polymers

L Cohen

Electronic Properties Of Conjugated Polymers:

This book delves into Electronic Properties Of Conjugated Polymers. Electronic Properties Of Conjugated Polymers is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Electronic Properties Of Conjugated Polymers, encompassing both the fundamentals and more intricate discussions.

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